

Chemistry Std 6

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ASSIGNMENT No. 19

Chapter: Matter

Q1. Define intermolecular space and intermolecular force. (Hint: refer Topic intermolecular space and intermolecular force)

Q2. Compare solids, liquids and gases on the basis of following properties: (i) shape (ii) volume (iii) compressibility (iv) position of particles (Hint: refer table 3.1)

Q3. Why intermolecular force in solid is maximum? (Hint: refer Topic intermolecular force)

Q4. Why intermolecular space in gas is maximum? (Hint: refer Topic intermolecular space)